

## Appendix A : $\Gamma$ -space and $\mu$ -space

III-A1

- In Section F, we introduced the  $\Gamma$ -space for the phase space of a  $N$ -particle system.

In 3D, we need to specify

$$\{ \underbrace{x_1, y_1, z_1, p_{1x}, p_{1y}, p_{1z}}_{\text{position and momentum of particle \#1}}, \dots, \underbrace{x_N, y_N, z_N, p_{Nx}, p_{Ny}, p_{Nz}}_{\text{position and momentum of particle \#N}} \}$$

6N quantities

$\Gamma$ -space : 6N dimensional space

One point in  $\Gamma$ -space specifies 6N quantities

$\Rightarrow$  One point in  $\Gamma$ -space specifies a state of the  $N$ -particle system.

III-A2

## $\mu$ -space

- This is the phase space for one particle.

For a particle in a 3D system, its phase space is a 6-dimensional space. The 6D axes are:

$$\{x, y, z, p_x, p_y, p_z\}$$

When a particle takes on  $\{x_1, y_1, z_1, p_{1x}, p_{1y}, p_{1z}\}$ , its state is represented by a point in  $\mu$ -space.

Thus, a particle's state  $\Rightarrow$  a point in  $\mu$ -space

A  $N$ -particle state  $\Rightarrow$  states of  $N$  particles

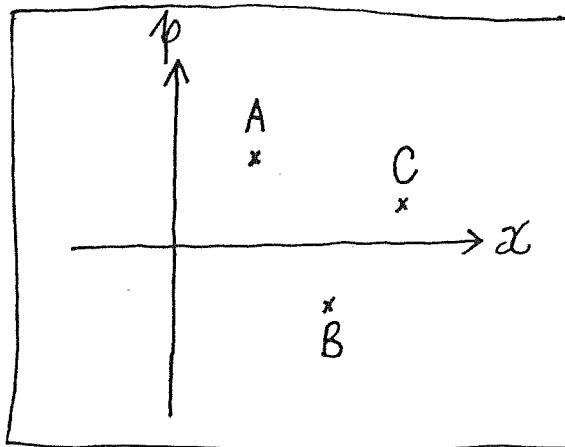
$\Rightarrow$   $N$  points in  $\mu$ -space

Ref:

Phase-space usually appears in Classical Mechanics under the Chapter on Hamilton's Dynamics. In particular, the related Liouville's theorem is often the starting point in the discussion on how a system evolves in time in Statistical Dynamics. I avoided the discussion so as not to confuse you. For applying Equilibrium Stat. Mech., we do not need that discussion.

- Example: 3 distinguishable particles in 1D

$\mu$ -space in 1D:  $\{x, p\}$  (2D  $\mu$ -space)



specifies the values  
of  $(x_A, p_A), (x_B, p_B), (x_C, p_C)$ ,  
⇒ specifies one state  
of the 3-particle system.

- Questions:

- how QM ideas could get into the picture?
- how distinguishability and indistinguishability of the particles get into the picture?
- When we are sure that there won't be two or more particles in the same " $(x, p)$ " state, how can we relate the counting when A, B, C are distinguishable to the counting when A, B, C are indistinguishable? [Correcting over-counting]  
( $\frac{1}{3!}$  or  $\frac{1}{N!}$  works!)

## Appendix B : Stirling's Approximation

- In stat. mech., we encounter  $\ln N!$  very often, where  $N$  is a high number (e.g.  $\sim 10^{23}$ )

Stirling's formula:  $n! \approx \sqrt{2\pi n} n^n e^{-n}$  (\*)

OR  $\ln N! \approx N \ln N - N + \frac{1}{2} \ln(2\pi N) + O(\frac{1}{N})$

e.g.  $N=50$ ,  $\ln N! = 148.478$  (exact)

$$\begin{aligned} N \ln N - N &= 145.601 \\ \frac{1}{2} \ln(2\pi N) &= 2.875 \end{aligned} \quad \left. \begin{aligned} N \ln N - N &> \frac{1}{2} \ln(2\pi N) \\ \text{for large } N \end{aligned} \right.$$

e.g.  $N=10^{18}$ ,

$$N \ln N - N = 4.0446 \times 10^{19}$$

$$\frac{1}{2} \ln(2\pi N) \approx 21.64 \text{ (tiny!)}$$

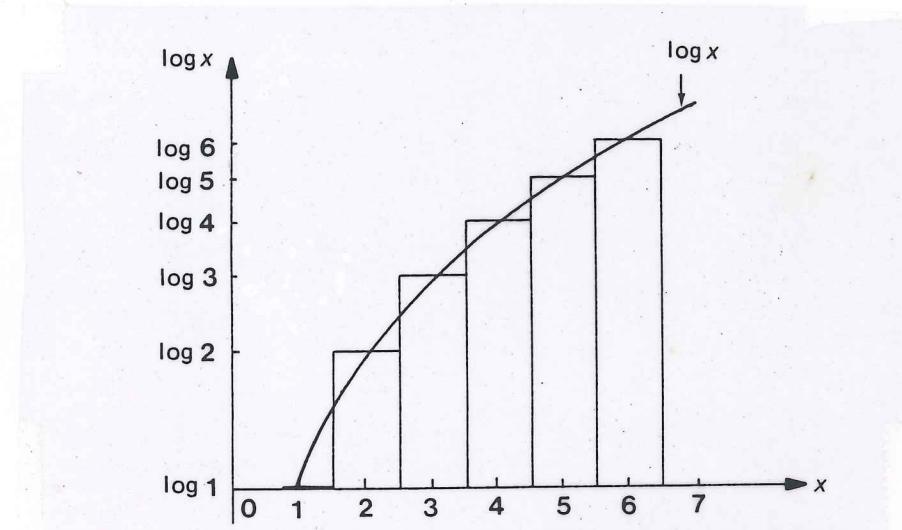
∴ Practically, for typical stat. mech. situations:

$\ln N! \approx N \ln N - N$       Stirling's formula

\* The proof is quite involved. It is related to the Gamma function, Laplace transform, and functions of a complex variable. For a version of the proof and how well the formula works, see Problem Set 1.

- A poorman's way of looking at the Stirling's formula.

$$\begin{aligned} \ln N! &= \ln[N(N-1)(N-2)\cdots 2 \cdot 1] \\ &= \sum_{i=1}^N \ln i \end{aligned}$$



$$\sum_{i=1}^N \ln i = \text{sum of areas under the rectangles}$$

- For  $N > 1$ , approximate areas under rectangular by area under the curve  $\ln x$ .

i.e.  $\sum_{i=1}^N \ln i \approx \int_1^N \ln x dx = N \ln N - N + \frac{1}{2} \approx N \ln N - N$